

Answers for class prep quiz on section 3.8, Stewart's Calculus (8th ed.)

1. **Answer:** (d). What is actually the case with  $T(t)$  is that if  $T_s$  is room temperature and we let  $y(t) = T(t) - T_s$ , then  $y(t) = Ce^{kt}$  for some constants  $C$  and  $k$ , or in other words,  $T(t) = T_s + Ce^{kt}$ .
2. **Answer:** (a). We know that  $P(t) = P_0e^{kt}$  ( $t$  in years), and we are given that  $P(1) = 20000$  and  $P(5) = 60000$ . In other words,

$$20000 = P_0e^k, \quad 60000 = P_0e^{5k}.$$

Dividing the second equation by the first, we get

$$3 = e^{5k}/e^k = e^{4k}.$$

Taking  $\ln$  of both sides, we see that  $\ln 3 = 4k$ , or  $k = \frac{\ln 3}{4}$ . Therefore, since  $20000 = P_0e^{(\ln 3)/4}$ ,  $P_0 \approx 15,200$ , and the population at  $t = 6$  is

$$P(6) = 15,000e^{6(\ln 3)/4} \approx 79000.$$

3. **Answer:** (a). Given  $m(t) = m_0e^{kt}$ , to say that the half-life is  $h$  means that  $m(h) = \frac{1}{2}m_0$ , which means that

$$\frac{1}{2}m_0 = m_0e^{kh},$$

and so  $e^{kh} = \frac{1}{2}$ , which means that  $k = \frac{\ln(1/2)}{h}$ . In fact, this formula is always true for radioactive decay, though the half-life  $h$  may vary.

4. **Answer:** (b). Let  $y(t) = T(t) - T_s = T(t) - 70$ . We know that  $y(t) = y_0e^{kt}$  for some constants  $y_0$  and  $k$ . Since  $T(0) = 98$ ,  $y_0 = y(0) = 28$ , and since  $T(1) = 93$ ,  $y(1) = 23$  and

$$23 = 28e^k,$$

which means that  $k = \ln(23/28)$ . Therefore,

$$T(3) = 70 + y(t) = 70 + 28e^{(\ln(23/28))3} \approx 85.5^\circ F.$$